# Carbonation in OPC and Alternative Cement Concrete: Impact of Real Climate versus Accelerated Testing and Implications on Service Life

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Exploratory Advanced Research (EAR)





#### What are ACMs?

Defining alternative binders (or alternative cements) has been an evolutionary process:

- Alternative (non-Portland) cements
- Per ACI ITG-10R-18:

alternative cement—an inorganic cement that can be used as a complete replacement for portland or blended hydraulic cements, and that is not covered by applicable specifications for portland or blended hydraulic cements.

ACI Committee
242: Alternative
Cements

#### What are ACMs?

# ACI ITG-10 classification of commercially available and emerging alternative cement technologies.

Clinkered alternative cements - ACM produced using technologies similar to portland cement production, with process changes that preclude production of portland cement but positively affect the environmental impact of production.

- Calcium aluminate (CAC)
- Reactive belite
- Calcium sulfoaluminate (CSA)
- Carbonated calcium silicate

**Calcined alternative cements -** ACM produced by calcining a raw material only, without further pyroprocessing, to produce additional mineral phases within the material.

- Magnesium oxychloride
- Magnesium phosphate (MPC)
- Magnesium ammonium phosphate
- Magnesium potassium phosphate

Nonclinkered alternative cements - ACM produced using precursors that require no pyroprocessing and set after addition of an activating solution to cause reactions that are not hydration or acid-base.

- Alkali-activated (AA)
- Fly ash
- Slag
- Recycled glass
- Supersulfated cement

#### **PROJECT OVERVIEW**

- FHWA Exploratory Advanced Research (EAR) project 2014-2019
- Novel Alternative Cementitious Materials for Development of the Next Generation of Sustainable Transportation Infrastructure



#### **Upscaling & Application**

- Approval process Adaptation of standards Performance criteria
- → variety of uses in transportation

Goal: Identify (and adapt) ACMs for rapid implementation relying upon existing construction technologies

#### **Durability Testing &** Advanced Characterization

- Unified approachAdaptation of standards
- Limits

# Preliminary Screening: Novel ACMs

- Range of chemistriesSustainability
- Past in situ performance
- Meet basic performance targets









#### **RESEARCH TEAM**





- Kimberly Kurtis
- Lisa Burris (now at Ohio State)
- **Prasanth Alapati**



- Tyler Ley
- Mehdi Khanzadeh
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Neal Berke

#### **US Army ERDC**

**US Army Corps** of Engineers®

- **Robert Moser**
- Sarah Williams













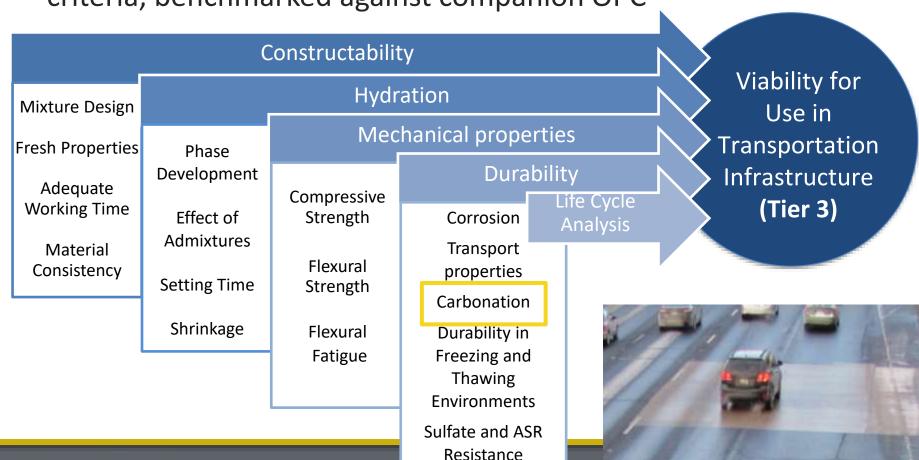




#### **PROJECT OVERVIEW**

- Phase 1: 12+ ACMs considered, 9 ACMs evaluated
- Phase 2, 3: 5 ACMs evaluated

 Many, many mix designs evaluated against performance criteria, benchmarked against companion OPC



# Why are we interested in ACMs?

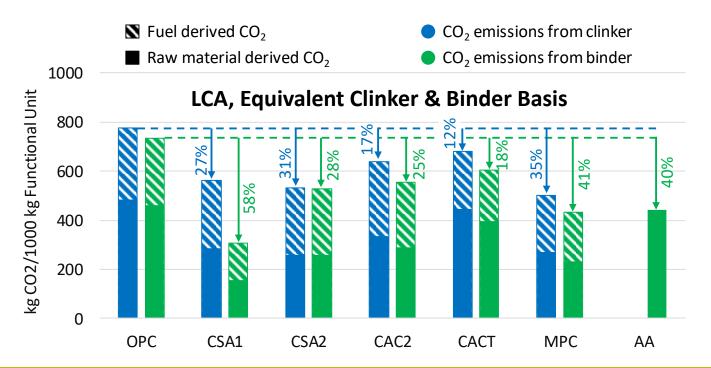
Unique properties appropriate for specialty applications could benefit larger scale transportation infrastructure construction

- Repair (fast set)
- Shrinkage compensation
- High early strength
- Low heat of hydration
- Enhanced durability
- New construction methods

# Why are we interested in ACMs?

#### Sustainable construction materials

- Lower embodied CO<sub>2</sub>
- Lower embodied energy
- Higher strength → smaller dimensions
- Enhanced durability → longer service life

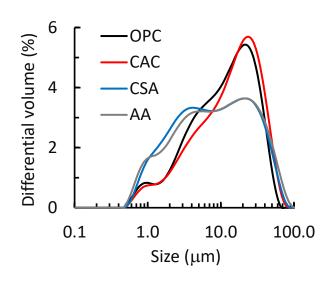


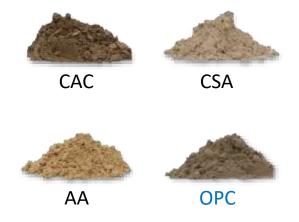
# Agenda

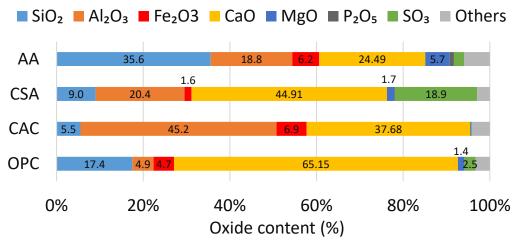
- Carbonation in ACM vs OPC concrete
  - Reaction mechanisms
  - Effects on porosity, performance
  - Carbonation rates
- Influence of CO<sub>2</sub> concentration
  - Atmospheric levels vs accelerated testing
- Conclusions & Recommendations

#### Materials: 3 ACMs (out of 9)

Binder		Specific gravity	SSA (m²/g)
Calcium Aluminate cement	CAC	2.97	306.4
Calcium SulfoAluminate cement	CSA	2.81	501.3
Alkali-Activated binder (C-ash)	AA	2.58	550.7
Type I/II Ordinary Portland cement	OPC	3.05	333.3

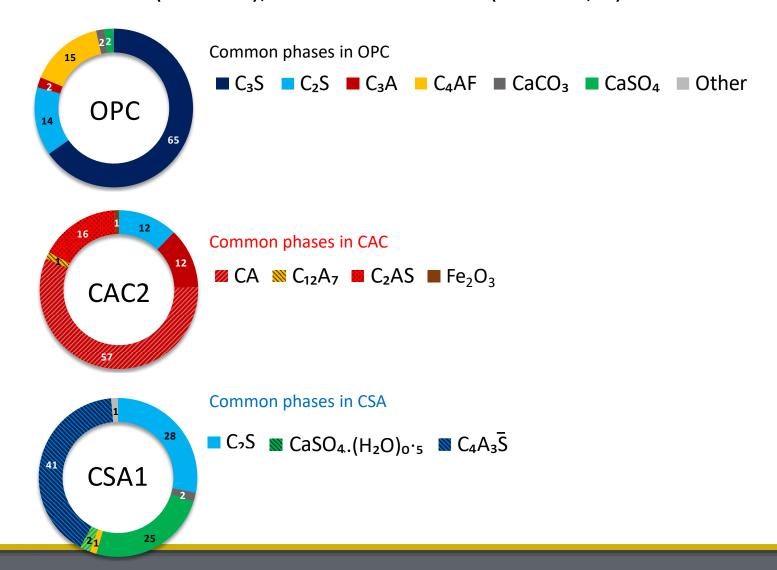






## **Varied Phase Composition**

9 ACMs evaluated (Phase 1); 5 ACMs evaluated (Phase 2, 3)



# **Different hydration products**

$$2C_3S + 6H \rightarrow \textbf{C-S-H} + 3CH$$
 OPC  
 $C_2S + 4H \rightarrow \textbf{C-S-H} + CH$   
 $C_3A + 3C\overline{S}H_2 + 26H \rightarrow \textbf{C}_6A\overline{S}_3H_{32}$  (Ettringite)  
 $2C_3A + C_6A\overline{S}_3H_{32} + 4H \rightarrow 3\textbf{C}_4A\overline{S}H_{12}$  (Afm)

2CA + 60H 
$$\rightarrow$$
  $2C_3AH_6 + 4AH_3 + 36H$ 

CSA
$$C_{4}A_{3}\overline{S} + 2C\overline{S} + 38H \rightarrow \textbf{Ettringite} + 2AH_{3}$$

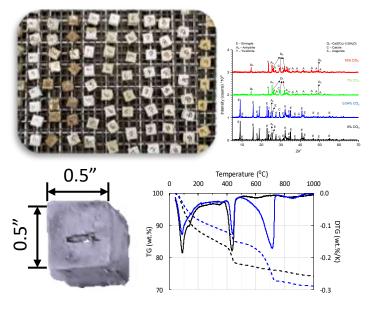
$$C_{4}A_{3}\overline{S} + 18H \rightarrow \textbf{Afm} + 2AH_{3}$$

$$C_{4}A_{3}\overline{S} + 8C\overline{S} + 6CH + 90H \rightarrow 3\textbf{Ettringite}$$

(Ca + Si) + H 
$$\rightarrow$$
 **C-S-H**  
(Si + Al) + Alkaline sol  $\rightarrow$  (alkali cation)-A-S-H

## **Understanding Carbonation in ACMs vs OPC: 3 Objectives**

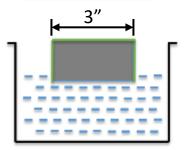
Reaction mechanisms
 (hydration and carbonation products)



Cement paste cubes at w/c: 0.45 (0.25-AA)

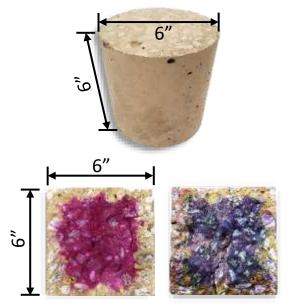
2. Effects on porosity





Cement mortar discs at w/c : 0.45 (0.25-AA)

3. Effects on pH, and rate of carbonation



Concrete cylinders at w/c : 0.4 (0.205–AA)

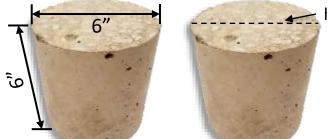
# **ACM Durability - Approach**

# Laboratory testing of ACM concrete durability

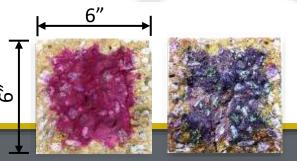
- Rely, as much as possible, on established accelerated test methods
- Compare against OPC
- Relate to materials characterization to understand factors governing performance
- Validate against field performance

Exposed to **7% CO<sub>2</sub>** at 30C and 55%RH for 56 days (84 days – concrete)

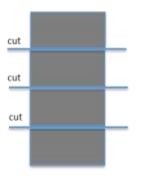




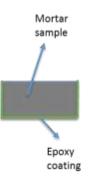
Split into two halves along this line

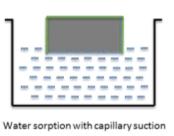


# **Sorptivity Test – ASTM C1585**

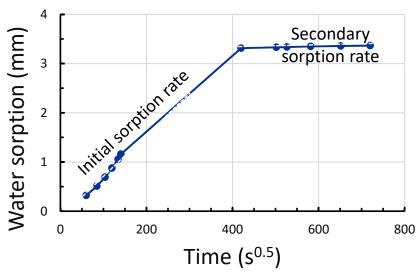






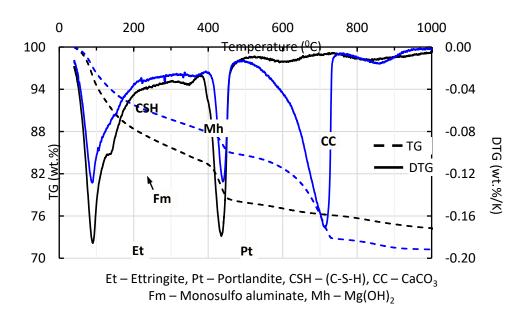






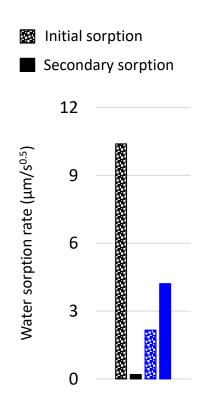
#### **Carbonation in OPC**

CO<sub>2</sub> exposure level: 0, 7% for 56 days



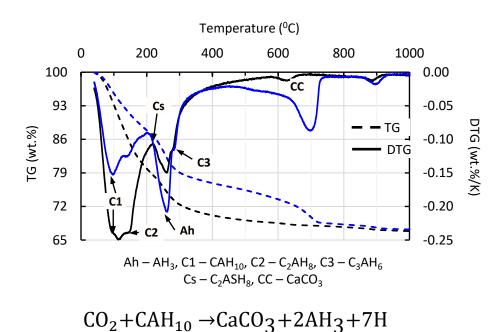
$$CH + CO_2 \rightarrow CaCO_3 + H$$

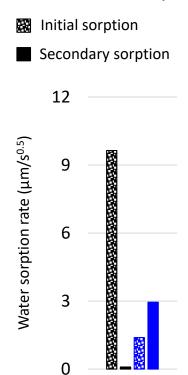
- Decomposition of secondary phases might not significantly affect the strength
- Initial sorption ↓, secondary sorption ↑
   Dissolution of phases (CH) into pore water -> precipitating as CaCO<sub>3</sub> in capillary pores
- CH buffers the pore solution --> carbonation might reduce the pH



#### **Carbonation in CAC**

CO<sub>2</sub> exposure level: 0, 7% for 56 days

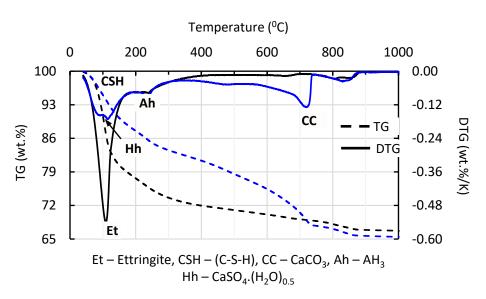




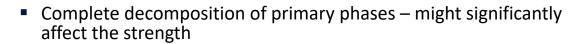
- Decomposition of primary phases might significantly affect the strength
- Initial sorption ↓, secondary sorption ↑
   Dissolution of CAH<sub>10</sub> phase into pore water -> precipitating as CaCO<sub>3</sub> in capillary pores
- CAH<sub>10</sub> buffers the pore solution --> carbonation might significantly reduce the pH

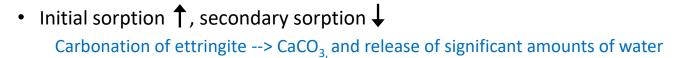
#### **Carbonation in CSA**

CO<sub>2</sub> exposure level: 0, 7% for 56 days

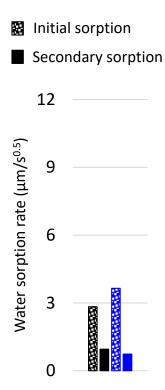


Ettringite 
$$+ CO_2 \rightarrow 3CaCO_3 + 3C\overline{S}H_x + AH_3 + (29-x)H$$



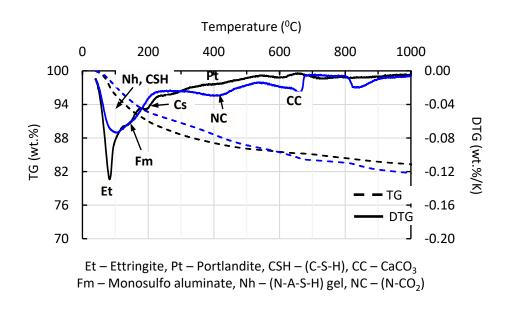


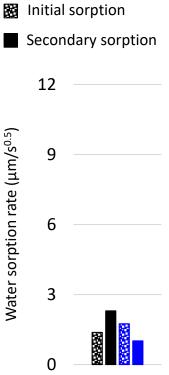
Acidification of pore water with CO<sub>2</sub>--> carbonation might reduce the pH



#### **Carbonation in AA**

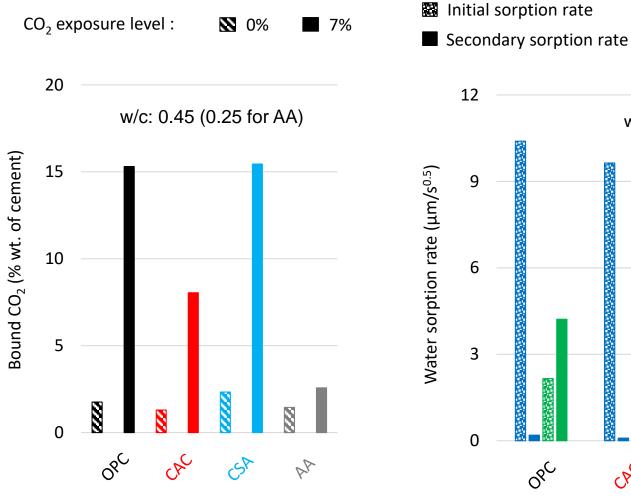
CO<sub>2</sub> exposure level: 0, 7% for 56 days

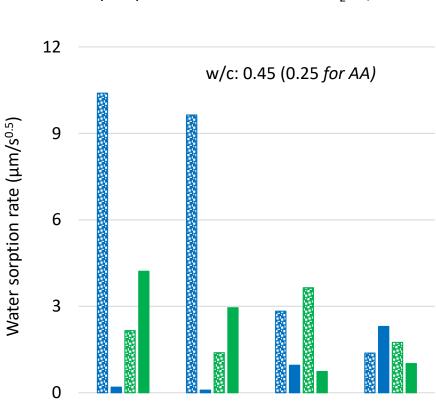




- Decomposition of secondary phases might not significantly affect the strength
- Initial sorption ↑, secondary sorption ↓
   Carbonation of ettringite -> CaCO<sub>3</sub>, and release of significant amounts of water
- Carbonation of alkalis --> might significantly reduce the pH

# CO<sub>2</sub> Binding, Effects on Porosity



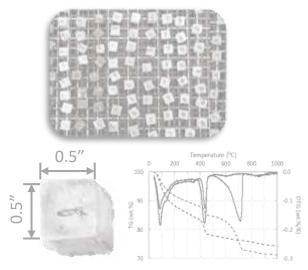


0% CO<sub>2</sub> Exposure

7% CO<sub>2</sub> Exposure

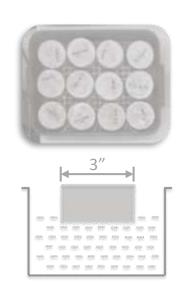
#### **Carbonation Rates for ACMs vs OPC**

Reaction mechanisms
 (hydration and carbonation products)

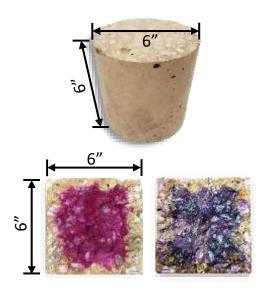


Cement paste cubes at w/c: 0.45 (0.25-AA)

2. Effects on porosity

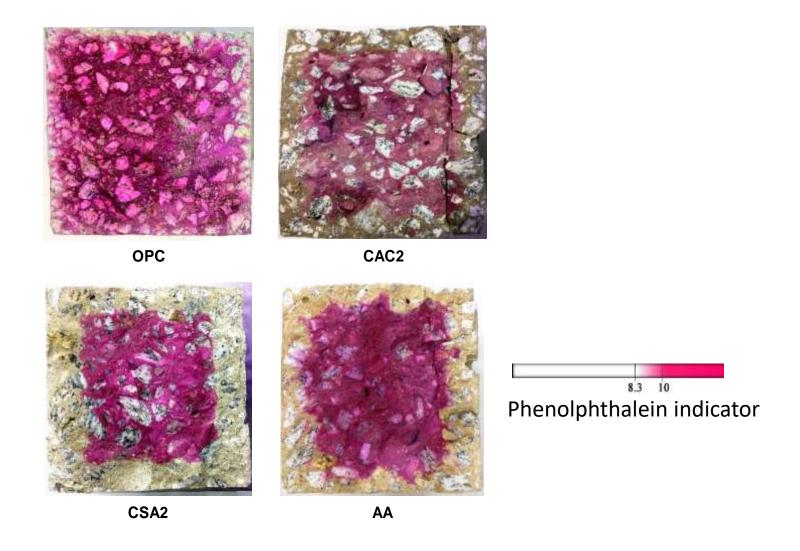


Cement mortar discs at w/c : 0.45 (0.25–AA) 3. Effects on pH, and rate of carbonation

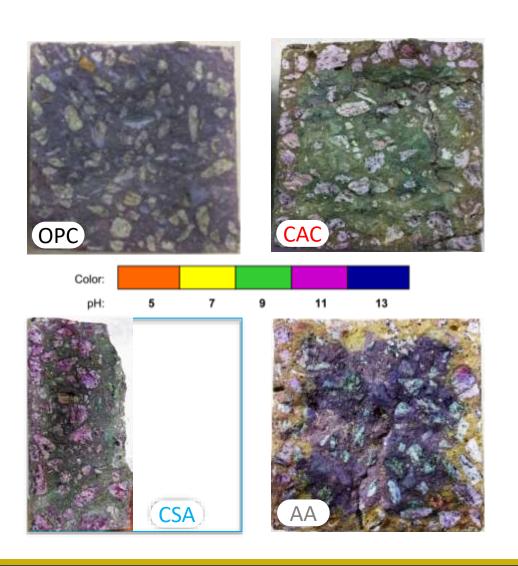


Concrete cylinders at w/c: 0.4 (0.205–AA)

## **Carbonation in Concrete**



# **Effects of carbonation on pH**

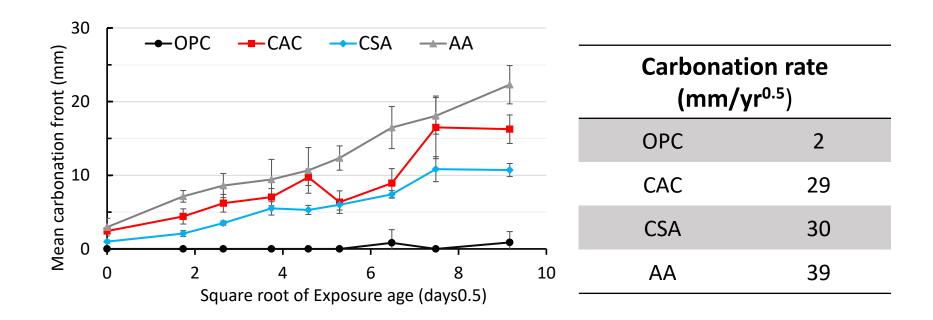


- Exposed to 7% CO<sub>2</sub> for **84 days** @
   30C & 55%RH
- Split into 2 halves -> sprayed with "rainbow" indicator

#### рΗ

	Uncarbonated	Carbonated
OPC	> 13	9 - 11
CAC	9 - 11	< 9
CSA	9 - 11	< 9
AA	> 13	< 9

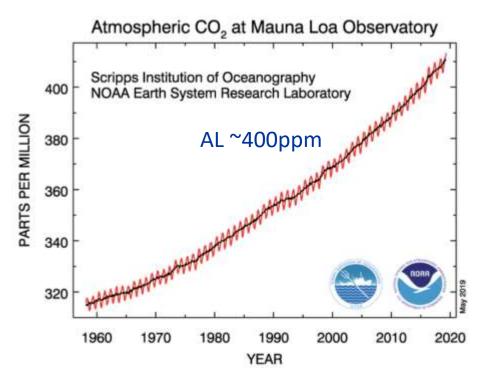
# Carbonation rate in ACMs compared to OPC



- Even though AA has low porosity and higher pH, the amount of CO<sub>2</sub> required to carbonate is significantly lower compared to CSA and OPC
- In CAC, reduction in capillary porosity with carbonation
- OPC has higher amounts of CH good buffering capacity (compared to ACMs) even at higher levels of carbonation

# Agenda

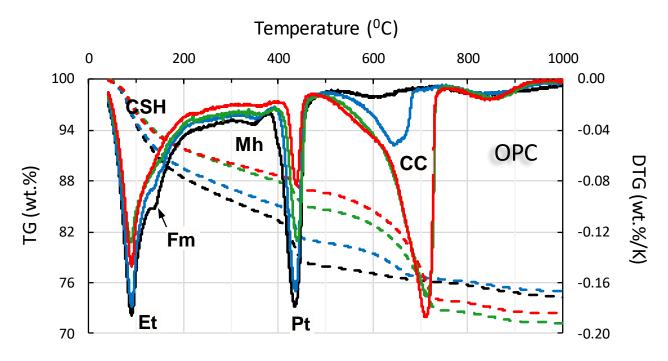
- Carbonation in ACM vs OPC concrete
  - Reaction mechanisms
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- Influence of CO<sub>2</sub> concentration
  - Atmospheric levels vs accelerated testing
- Conclusions & Recommendations



# OPC: Effect of [CO<sub>2</sub>]

CO<sub>2</sub> exposure level: **0**, **0.04**, **7**, **15%** for 56 days

— TG — DTG



Et – Ettringite, Pt – Portlandite, CSH – (C-S-H), CC – CaCO<sub>3</sub> Fm – Monosulfo aluminate, Mh – Mg(OH)<sub>2</sub>

 $CH + CO_2 \rightarrow CaCO_3 + H$ 

#### **Conclusions: ACM vs OPC Carbonation**

- CAC, CSA and AA concretes carbonated at a much faster rate compared to OPC concrete.
- Carbonation in CAC and CSA --> significant decomposition of main hydration products.
- Carbonation in OPC and CAC mortar mixtures resulted in significant reduction in near-surface capillary porosity.
- pH levels in the carbonated region of ACM concrete dropped below 9.
  - Effects on steel passivation and corrosion must be examined.

More detailed presentation of these results: Alapati and Kurtis, Sixth International Conference on the Durability of Concrete Structures, 18-20 July 2018, University of Leeds.

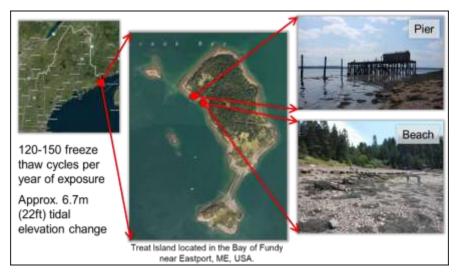
# **Conclusions: Carbonation Testing**

- Increased [CO<sub>2</sub>] produced more rapid rates of carbonation, but with mechanisms relatively consistent with exposure to atmospheric levels, except for AA
  - → Higher concentrations likely not appropriate for AA systems
  - → Carbonation appears to facilitate conversion
  - → Relevant for test method design
  - → Provides some insight into potential effects of increasing global GHG emissions
- Need to establish consistent accelerated exposure test, considering sample preparation (w/c, curing, age at testing) and exposure (duration, CO<sub>2</sub> concentration, temperature, RH)
- Correlation with other performance metrics (e.g.,sorptivity, diffusivity, corrosion onset, compressive strength) necessary
- Validation against field performance necessary

# **Field Durability Studies**

- Treat Island, Maine: wet/dry, freeze/thaw, corrosion
- Biscayne Bay, Florida: wet/dry, corrosion
- Yearly monitoring during site inspections:
  - Visual and NDE
- Retrieval for lab testing:
  - Petrographic analysis
  - Chloride diffusion
  - Mechanical properties





Treat Island – Coastal Maine Exposure



Biscayne Bay – Coastal Florida Exposure

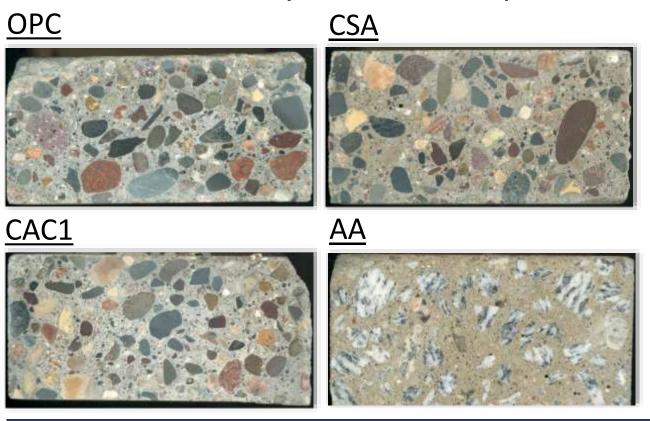
# **Field Durability Studies**

Photos after 3yr of severe exposure at Treat Island:

**OPC CSA** CAC

# **Field Durability Studies**

Petro after 2yr of severe exposure at Treat Island:



ACMs have faired remarkably well in this severe environment, with as good or better performance than OPC concrete – no freeze/thaw damage, no scaling, edge retention – NO DAMAGE

#### **Recommendations: Carbonation**

- Carbonation may be a concern for all ACMs used in lowcover reinforced concrete subject to moderate humidity conditions.
  - Of all ACMs considered, a blend of CAC and OPC performed best (not presented here)

OPC out-performed all ACMs, including CAC/OPC blend

 Further studies are needed to assess the influence of carbonation on passivation behavior in reinforced ACM concrete.

Electrochemical tests on carbonated systems are being performed

### **Questions?**

#### **Discussion**

Acknowledgement:

